1



Alkenes

Specification reference	Checklist questions	
4.1.3 a	Can you describe alkenes as unsaturated hydrocarbons containing a C=C bond comprising a π -bond (sideways overlap of adjacent p-orbitals above and below the bonding C atoms) and a σ -bond (overlap of orbitals directly between the bonding atoms); with restricted rotation of the π -bond?	
4.1.3 b	Can you explain the trigonal planar shape and bond angle around each carbon in the C=C of alkenes in terms of electron pair repulsion?	
4.1.3 c i	Can you explain the term stereoisomers (compounds with the same structural formula but with a different arrangement in space)?	
4.1.3 c i	Can you explain the term <i>E/Z isomerism</i> (an example of stereoisomerism, in terms of restricted rotation about a double bond and the requirement for two different groups to be attached to each carbon atom of the C=C group)?	
4.1.3 c i	Can you explain the term <i>cis–trans isomerism</i> (a special case of <i>E/Z</i> isomerism in which two of the substituent groups attached to each carbon atom of the C=C group are the same)?	
4.1.3 c ii	Can you use Cahn–Ingold–Prelog (CIP) priority rules to identify the <i>E</i> and <i>Z</i> stereoisomers?	
4.1.3 d	Can you determine possible <i>E/Z</i> or <i>cis–trans</i> stereoisomers of an organic molecule, given its structural formula?	
4.1.3 e	Can you describe the reactivity of alkenes in terms of the relatively low bond enthalpy of the π -bond?	
4.1.3 f i	Can you describe addition reactions of alkenes with: hydrogen in the presence of a suitable catalyst (e.g. Ni) to form alkanes?	
4.1.3 f ii	Can you describe addition reactions of alkenes with halogens to form dihaloalkanes, including the use of bromine to detect the presence of a double C=C bond as a test for unsaturation in a carbon chain?	
4.1.3 f iii	Can you describe addition reactions of alkenes with hydrogen halides to form haloalkanes?	



Specification reference	Checklist questions	
4.1.3 f iv	Can you describe addition reactions of alkenes with steam in the presence of an acid catalyst (e.g. H ₃ PO ₄) to form alcohols?	
4.1.3 g	Can you define and use the term <i>electrophile</i> (an electron pair acceptor)?	
4.1.3 h	Can you explain the mechanism of electrophilic addition in alkenes by heterolytic fission?	
4.1.3 i	Can you use Markownikoff's rule to predict formation of a major organic product in addition reactions of H–X to unsymmetrical alkenes (e.g. H–Br to propene) in terms of the relative stabilities of carbocation intermediates in the mechanism?	
4.1.3 j i	Can you describe addition polymerisation of alkenes and substituted alkenes, including the repeat unit of an addition polymer deduced from a given monomer?	
4.1.3 j ii	Can you describe addition polymerisation of alkenes and substituted alkenes, including identification of the monomer that would produce a given section of an addition polymer?	
4.1.3 k i	Can you list and describe the benefits for sustainability of processing waste polymers by combustion for energy production?	
4.1.3 k ii	Can you list and describe the benefits for sustainability of processing waste polymers by use as an organic feedstock for the production of plastics and other organic chemicals?	
4.1.3 k iii	Can you list and describe the benefits for sustainability of processing waste polymers by removal of toxic waste products formed during disposal by combustion of halogenated plastics (e.g. PVC)?	
4.1.3 l	Can you list and describe benefits to the environment of development of biodegradable and photodegradable polymers?	